# Science—Term 6

Year 7 Term 6





### (1) Biology—Respiration

A chemical reaction that takes place in the mito- chondria. Releases energy for life processes.

#### **Aerobic Respiration:**

oxygen + glucose → water + carbon dioxide

#### **Anaerobic Respiration: In plants and animals**

glucose → lactic acid

Additional oxygen is needed to get rid of lactic acid.

The oxygen required to do this is called the oxygen debt.

#### **Anaerobic Respiration in Fungi (fermentation)**

glucose -> ethanol + carbon dioxide

Anaerobic respiration releases less energy than aerobic respiration

## (2) Chemistry— Investigating Catalysts

#### **Catalysts**

A substance which increases the rate of a chemical reaction without being used up.

Accurate	The result is close to the expected value
Precise	All of the results are close together
Repeatable	The same results can be achieved when the experiment is repeated
Method	A clear and concise set of instructions to carry out an investigation

(4) Chemistry—Exothermic and Endothermic Reactions

A reaction which releases energy in the form of heat to the surroundings.

### (3) Physics— Energy Sources

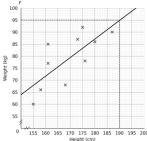
Renewable	An energy source that <u>will not</u> run out on a time scale of 1 million years e.g. solar and wind.
Non-renewable	An energy source that <u>will</u> run out on a time scale of 1 million years e.g. coal, oil or gas.



## (4) Biology— Investigating Effect of Exercise

Independent Variable	The factor that is changed in the investigation. Found in the first column of a table, and the x axis of a graph.
Dependent Variable	The factor that is measured in the investigation. Found in the y axis of a graph.
Control Variable	Factors in the investigation that must be kept the same for each repeat.

#### Presenting data-line graph



- Independent variable on the x axis
- Dependent variable on the y axis
- Both axis labelled with units
- Line of best fit plotted

"As the (x axis value) increases the (y axis value) .....

### **Endothermic**

**Exothermic** 

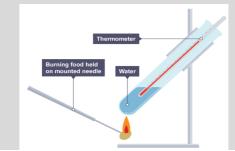
A reaction which absorbs energy in the form of heat from the surroundings.

## (6) Physics — Energy

Energy	A quantity required to do work e.g. movement or heating.
Power	The rate of energy transfer, measured in watts (W).  Power (W) = Energy (J) ÷ time (S)

#### **Converting units:**

**Energy in Food Investigation** 1000 W = 1 kW **<u>Hazard:</u>** something that could cause harm 1000 J = 1 kJ to you or another person e.g. broken glass 3,600,000 J = 1 kWh Risk: how the hazard causes harm e.g. cuts 3600 s = 1 h



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